

Synthesis of Unnatural Amino Acids via Suzuki Cross-Coupling of Enantiopure Vinyloxazolidine Derivatives

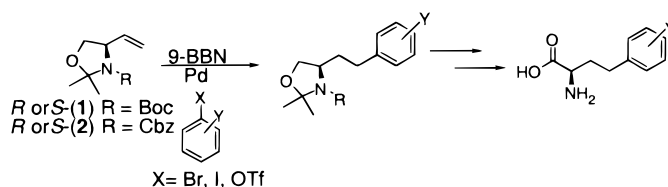
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ABSTRACT



(*R* and *S*)- α -Amino alcohols and α -amino acids, including 4-methoxyhomophenylalanine, with a variety of unnatural side chains have been synthesized via palladium-catalyzed cross-coupling Suzuki reactions. The key building blocks **1** and **2**, synthesized from the common achiral precursor 2-butene-1,4-diol, were made enantiopure utilizing a *Pseudomonas cepacia* lipase-catalyzed kinetic resolution. The optimal conditions for the Suzuki cross-coupling and the subsequent oxidations of the resultant α -amino alcohols are described.

Nonproteinogenic α -amino acids have been widely used as synthetic building blocks and display interesting biological properties.¹ Analogues of homophenylalanines,² such as 4-methoxyhomophenylalanine (**40**),^{2e,3} have received particular attention as constituents of potential pharmaceuticals. A methodology to the synthesis of α -amino acids and α -amino alcohols, of both *R* and *S* epimers and containing a variety of unnatural side chains was desired. Recently Taylor and co-workers published the synthesis of unnatural α -amino acids utilizing palladium-catalyzed Suzuki cross-coupling.⁴ Similar work has been the focus of this laboratory; however, the protocol disclosed in this Letter offers significant advantages in terms of coupling efficiency and diver-

sity: specifically, the synthesis and use of (*R*)- and (*S*)-**1** and **-2**, the development of cross-coupling methodology compatible with aryl triflates and nitrogen heterocycles, and facile oxidation protocols.

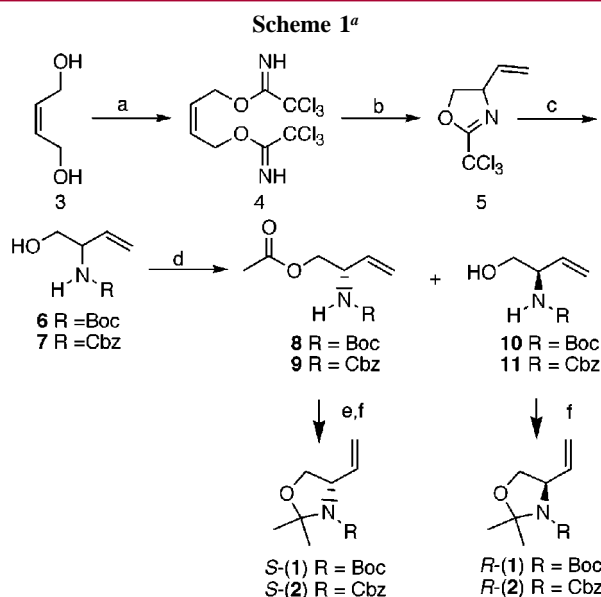
Compounds **1**⁵ and **2**⁶ have previously been prepared and exploited as versatile intermediates. The Suzuki cross-coupling reaction of these compounds would constitute an efficient synthesis of unnatural α -amino acids through direct introduction of functional groups to the amino acid moiety. Our synthesis of (*R*)- and (*S*)-**1** and (*R*)- and (*S*)-**2** is illustrated in Scheme 1. Commercial grade *cis*-butene-1,4-diol (**3**) was converted to bis-imidate **4**.⁷ Treatment of this material with PdCl₂(CH₃CN)₂ afforded the unexpected oxazoline **5** presumably via a π -allyl mechanism.⁸ Hydrolysis

(1) (a) Sardina, F. J.; Rapoport, H. *Chem. Rev.* **1996**, *96*, 1825. (b) Duthaler, R. O. *Tetrahedron* **1994**, *50*, 1539. (c) Williams, R. M. *Synthesis of Optically Active α -Amino Acids*; Baldwin, J. E., Ed.; Pergamon Press: Oxford, 1989. (d) Barrett, G. C. *Chemistry and Biochemistry of the Amino Acids*; Chapman and Hall: London, 1985.

(2) (a) Ehrlich, P. P.; Jeffrey, W. R.; Michaelides, R. M. *J. Org. Chem.* **1997**, *62*, 2782. (b) Petasis, N. A.; Zavialov, I. A. *J. Am. Chem. Soc.* **1997**, *119*, 445. (c) Baxter, A. D.; Murray, P. J.; Taylor, R. J. H. *Tetrahedron Lett.* **1992**, *33*, 2331. (d) Jackson, R. F. W.; Wishart, N.; Wood, A.; James, K.; Wythes, M. J. *J. Org. Chem.* **1992**, *57*, 3397. (e) Melillo, D. G.; Larsen, R. D.; Mathre, D. J.; Shukis, W. F.; Wood, A. W.; Colletuori, J. R. *J. Org. Chem.* **1987**, *52*, 5143. (f) Nordlander, J. E.; Payne, M. J.; Njoroge, F. G.; Visahwanath, V. M.; Gi, R. H.; Laikos, G. D.; Balk, M. A. *J. Org. Chem.* **1985**, *50*, 3619. (g) Urbach, H.; Henning, R. *Tetrahedron Lett.* **1984**, *25*, 1143. (h) Weller, H. N.; Gordon, E. M. *J. Org. Chem.* **1982**, *47*, 4160.

(3) (a) Yamada M.; Nagashima N.; Hasegawa J.; Takahashi S. *Tetrahedron Lett.* **1998**, *39*, 9019. (b) Cecchi, R.; Crochi, T.; Boge grain, R.; Boveri, S.; Baroni, M.; Boccardi, G.; Guimbard, J. P.; Guzzi, U. *Eur. J. Med. Chem.* **1994**, *29*, 259. (c) Williams, R. M.; Sinclair, P. J.; Zhai, D.; Chen, D. *J. Am. Chem. Soc.* **1988**, *110*, 1547. (d) Yato, M.; Homma, K.; Ishida, A. *Heterocycles* **1995**, *41*, 17. (e) Kosui, N.; Waki, M.; Kato, T.; Izumiya, N. *Bull. Chem. Soc. Jpn.* **1982**, *55*, 918. (f) Evans, W. C.; Walker, N. *J. Chem. Soc.* **1947**, 1571. (g) Sahoo, S. P.; Caldwell, C. G.; Chapman, K. T.; Durette, P. L.; Esser, C. K.; Kopka, I. E.; Polo, S. A.; Sperow, K. M.; Niedzwiecki, L. M.; Izquierdo-Martin, M.; Chang, B. C.; Harrison, R. K.; Stein, R. L.; MacCoss, M.; Hagmann, W. K. *Bioorg. Med. Chem. Lett.* **1995**, *5*, 2441.

(4) Campbell A. D.; Raynham T. M.; Taylor R. J. K. *Tetrahedron Lett.* **1999**, *40*, 5263.



^a (a) CCl_3CN , KH , 90–93%; (b) $\text{PdCl}_2(\text{CH}_3\text{CN})_2$, THF , 85%; (c) i. 6 N HCl ; ii. Boc_2O (**6**), Cbz_2O (**7**), 55%–65% (two steps); (d) PS-30 lipase, isopropenyl acetate, 38–46% (94%–99% ee); (e) KCN , 93% MeOH ; (f) DMP, acetone, $\text{BF}_3 \cdot \text{Et}_2\text{O}$, quantitative.

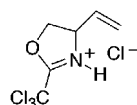
of the somewhat volatile **5** proceeded best under acidic conditions. Subsequent protection of the resultant amine salt with either Boc_2O or Cbz_2O ¹⁰ under biphasic conditions afforded compounds **6**¹¹ or **7**, respectively.¹² Kinetic resolutions of these compounds were achieved with *Pseudomonas cepacia* (Amano PS-30) lipase in CH_2Cl_2 /isopropenyl acetate. For the *N*-Boc protected **6**, the kinetic resolution was stopped at 50% completion to afford **10** in 46% chemical yield with $[\alpha]^{20}_{\text{D}} -28.5$ (*c* 1.0, CHCl_3) [lit.⁵ⁱ $[\alpha]^{25}_{\text{D}} -30.5$ (*c* 1.2, CHCl_3)]. Treatment of **8** with KCN followed by 2,2-

(5) (a) Beaulieu, P. L.; Duceppe, J.; Johnson, C. *J. Org. Chem.* **1991**, *56*, 4196. (b) Shimamoto, K.; Ishida, M.; Shinozaki, H.; Ohfuné, Y. *J. Org. Chem.* **1991**, *56*, 4167. (c) Boyd, E. C.; Paton, R. M. *Tetrahedron Lett.* **1993**, *34*, 3169. (d) McKillop, A.; Taylor, R. J. K.; Watson, R. J.; Lewis, N. *Synthesis* **1994**, 31. (e) Wei, Z.; Knaus, E. E. *Synthesis* **1994**, 1463. (f) Kawate, T.; Fukuta, N.; Nishida, A.; Nakagawa, M. *Chem. Pharm. Bull.* **1997**, *45*, 2116. (g) Moriwake, T.; Hamano, S.; Saito, S.; Torii, S. *Chem. Lett.* **1987**, 2085. (h) Kumar, J. S. R.; Datta, A. *Tetrahedron Lett.* **1997**, *38*, 6779. (i) Campbell, A. D.; Raynham, T. M.; Taylor, R. J. K. *Synthesis* **1998**, 1707.

(6) Reginato, G.; Mordini, A.; Capperucci, A.; Degl'Innocenti, A.; Manganiello, S. *Tetrahedron* **1998**, *54*, 10217.

(7) (a) Vyas, D. M.; Chiang, Y.; Doyle, T. W. *J. Org. Chem.* **1984**, *49*, 2037. (b) Cardillo, G.; Orena, M.; Sandri, S. *J. Org. Chem.* **1986**, *51*, 713.

(8) In our hands Overman rearrangement of the bisimidate **4** under reported conditions^{7a} resulted in incomplete conversion of starting material. Under more forceful thermal rearrangement conditions, significant losses were incurred due to charring and the formation of **5**· HCl



which sublimed at 180 °C. The structure of this material was verified by treatment of **5** with a dry ether solution of HCl which yielded a crystalline product with identical spectroscopic properties. Overman rearrangement (heating, K_2CO_3 to neutralize acidic species)⁹ likewise resulted in only partial conversion of starting material **4**.

(9) Nishikawa, T.; Asai, M.; Ohyabu, N.; Isobe, M. *J. Org. Chem.* **1998**, *63*, 188.

(10) Sennyey, G.; Barcelo, G.; Senet, J. *Tetrahedron Lett.* **1986**, *27*, 5375.

dimethoxypropane (DMP) afforded *S*-(**1**), $[\alpha]^{20}_{\text{D}} -17.0$ (*c* 1.4, CHCl_3) [lit.⁵ⁱ $[\alpha]^{20}_{\text{D}} -17.1$ (*c* 1.2, CHCl_3)]. Similar treatment of **10** with DMP resulted in (*R*)-**1**, $[\alpha]^{20}_{\text{D}} +17.2$ (*c* 1.2, CHCl_3). The kinetic resolution of the *N*-Cbz protected **7** proved more difficult and required termination at 40% conversion to obtain **11**, or at 60% conversion to obtain **9** with high enantioenrichment. Compound **11** displayed $[\alpha]^{20}_{\text{D}} -32.3$ (*c* 1.0, CHCl_3) [lit.^{12g} $[\alpha]^{25}_{\text{D}} -32.1$ (*c* 3.1, CHCl_3)]. Treatment of **9** with KCN resulted in the epimer of **11** with $[\alpha]^{20}_{\text{D}} +31.2$ (*c* 1.0, CHCl_3). The analogous (*S*)-**2** and (*R*)-**2** were obtained by similar treatment with DMP.

With the key building blocks **1** and **2** in hand, we embarked on the investigation of conditions for hydroboration of **1** and coupling with *p*-bromoanisole (Table 1) en

Table 1. Hydroboration Using 9-BBN and Suzuki Cross-Coupling Studies of Compound **1** with **12**¹³

| entry | hydroboration conditions | Pd cat. | coupling conditions | yield of 12a (%) |
|-------|--------------------------|----------|---------------------------------------|-------------------------|
| 1 | THF, rt | <i>a</i> | Cs_2CO_3 , rt | 39 |
| 2 | THF, rt | <i>a</i> | K_2CO_3 , DMF, rt | 25 |
| 3 | THF, 67 °C | <i>a</i> | K_3PO_4 , DMF, rt | 29 |
| 4 | THF, rt | <i>a</i> | 3.2 N NaOH , 55 °C | 66 |
| 5 | THF, 67 °C | <i>b</i> | 3.2 N NaOH , 80 °C | 72 |
| 6 | toluene, 80 °C | <i>b</i> | 3.2 N NaOH , 99 °C | 94 |
| 7 | toluene, 67 °C | <i>b</i> | CsF , 88 °C | 78 |
| 8 | toluene, 80 °C | <i>b</i> | 3 N NaOH , 80 °C | 80 |
| 9 | toluene, 80 °C | <i>b</i> | K_3PO_4 , DMF, 100 °C | 75 |
| 10 | toluene, 80 °C | <i>b</i> | K_2CO_3 , DMF, 100 °C | 56 |

^a 5 mol % of $\text{PdCl}_2(\text{dppf})_2$. ^b 3 mol % of $\text{Pd}(\text{PPh}_3)_4$.

route to 4-methoxyhomophenylalanine (**40**). Hydroboration of **1** in THF at rt proved to be exceedingly slow, and considerable starting material was present even after 24 h. However, when the reaction was run in toluene at 80 °C, the starting material disappeared completely within 15–30 min (entry 6).

With this satisfactory result, a wide variety of cross-coupling conditions were explored (Table 1). Our experiments indicate that a biphasic toluene/3.2 N NaOH system

(11) (a) Horikawa, M.; Hashimoto, K.; Shirahama, H. *Tetrahedron Lett.* **1993**, *34*, 331. (b) Kurokawa, N.; Ohfuné, Y. *Tetrahedron* **1993**, *49*, 6195. (c) Ohfuné, Y.; Kurokawa, N. *Tetrahedron Lett.* **1984**, *25*, 1071. (d) Crisp, G. T.; Gebauer, M. G. *Tetrahedron* **1996**, *52*, 12465. (e) Naito, T.; Ikai, M.; Shirakawa, M.; Fujimoto, K.; Ninomiya, I.; Kiguchi, T. *J. Chem. Soc., Perkin Trans. 1* **1994**, *7*, 773. (f) Ohfuné, Y.; Kurokawa, N. *Tetrahedron Lett.* **1984**, *25*, 1587. (g) Matsunaga, H.; Ishizuka, T.; Kunieda, T. *Tetrahedron* **1997**, *53*, 1275. (h) Kiguchi, T.; Ikai, M.; Shirakawa, M.; Fujimoto, K.; Ninomiya, I.; Naito, T. *J. Chem. Soc., Perkin Trans. 1* **1998**, *5*, 893. (i) Naito, T.; Shirakawa, M.; Ikai, M.; Ninomiya, I.; Kiguchi, T. *Recl. Trav. Chim. Pays-Bas* **1996**, *115*, 13. (j) See also refs 5a, 5b, 5c, 5d, 5g, and 5i.

(12) (a) Hayashi, T.; Yamamoto, A.; Ito, Y. *Tetrahedron Lett.* **1988**, *29*, 99. (b) Huwe, C. M.; Blechert, S. *Tetrahedron Lett.* **1995**, *36*, 1621. (c) Yoo, S.; Lee, S.; Jeong, N.; Cho, I. *Tetrahedron Lett.* **1993**, *34*, 3435. (d) Yoo, S.; Lee, S. *J. Org. Chem.* **1994**, *59*, 6968. (e) Wade, P. A.; Singh, S. M.; Pillay, M. K. *Tetrahedron* **1984**, *40*, 601. (f) Crisp, G. T.; Glink, P. T. *Tetrahedron* **1992**, *48*, 3541. (g) Trost, B. M.; Bunt, R. C. *Angew. Chem., Int. Ed. Engl.* **1996**, *35*, 99. (h) See also ref 11d. (i) Huwe, C. M.; Blechert, S. *Tetrahedron Lett.* **1994**, *35*, 9533; 9537. (j) Huwe, C. M.; Velder, J.; Blechert, S. *Angew. Chem., Int. Ed. Engl.* **1996**, *35*, 2376. (k) Huwe, C. M.; Blechert, S. *Synthesis* **1997**, 61. (l) See also refs 11c and 11f.

Table 2. Suzuki Coupling of (*R* and *S*)-3-(*tert*-Butoxycarbonyl)-2,2-dimethyl-4-vinylloxazolidine

| entry | coupling partner | product | yield (%) | entry | coupling partner | product | yield (%) |
|-------|------------------|---------|-----------------|-------|------------------|---------|-----------------|
| 1 | | | 94 | 10 | | | 67 |
| 2 | | | 85 | 11 | | | 36 |
| 3 | | | 87 | 12 | | | 78 ^b |
| 4 | | | 85 | 13 | | | 66 |
| 5 | | | 82 | 14 | | | 64 |
| 6 | | | 77 | 15 | | | 63 |
| 7 | | | 75 | 16 | | | 61 |
| 8 | | | 74 | | | | |
| 9 | | | 72 ^a | | | | |

^a PPTS in MeOH. ^b Chromatographically separable 1:1 mixture.

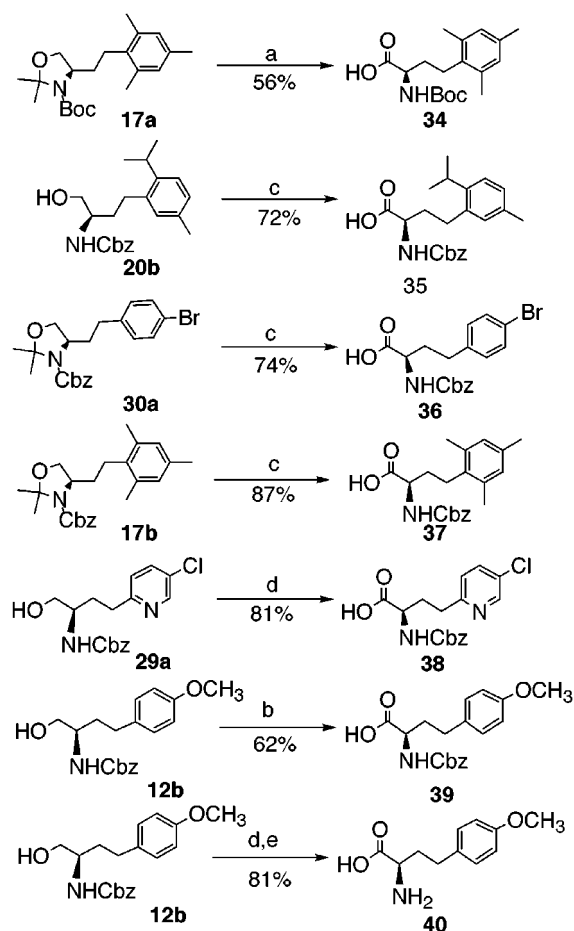
at elevated temperature and with Pd(PPh₃)₄ as catalyst resulted in maximum yields. Under these reaction conditions a variety of aryl halides and triflates including nitrogen heterocycles^{13c} underwent Suzuki cross-coupling with good to excellent yields (Table 2).

With a variety of protected chiral amino alcohol derivatives in hand, ways to convert these to amino acids were sought. Compounds **12a** and **17a** were chosen as representative examples (Scheme 2). The isopropylidene moiety was first removed using PPTS. Subsequent oxidation with various agents including PDC,^{14a} Jones,^{14b} O₂/Pt,^{14c} TPAP,^{14d} and CrO₃/H₅IO₆^{14e} gave either no reaction or a complex mixture of products. TEMPO^{14h} proved to be a capricious oxidant

which afforded no reaction in acetone¹⁴ⁱ but delivered the aldehyde smoothly in toluene/H₂O;^{14g} subsequent oxidation by NaClO₂ resulted in a modest yield of amino acid. A two-step procedure consisting of Sharpless RuCl₃ oxidation¹⁴ⁱ to the aldehyde stage followed with NaClO₂ treatment to obtain the amino acid **34** gave the best result.^{14f} However, even these conditions resulted in modest yields and were not general. As a more robust system was deemed necessary, the *N*-Cbz-protected analogues (Table 3) were examined. Compounds

(13) (a) Miyaura, N.; Suzuki, A. *Chem. Rev.* **1995**, *95*, 2457. (b) Miyaura, N.; Ishiyama, T.; Sasaki, H.; Ishikawa, M.; Satoh, M.; Suzuki, A. *J. Am. Chem. Soc.* **1989**, *111*, 314. (c) Watanabe, T.; Miyaura, N.; Suzuki, A. *Synlett* **1992**, 207. (d) Ishikura, M.; Kamada, M.; Terashima, M. *Synthesis* **1984**, 936. (e) Wright, S. W.; Hageman, D. L.; McClure, L. D. *J. Org. Chem.* **1994**, *59*, 6095.

(14) (a) Matsunaga, H.; Ishizuka, T.; Kunieda, T. *Tetrahedron* **1997**, *53*, 1275. (b) Dondoni, A.; Marra, A.; Massi, A. *Chem. Commun.* **1998**, 1741. (c) Maurer, P. J.; Takahata, H.; Rapoport, H. *J. Am. Chem. Soc.* **1984**, *106*, 1095. (d) Ley, S. V.; Norman, J.; Griffith, W. P.; Marsden, S. P. *Synthesis* **1994**, 639. (e) Zhao, M.; Li, J.; Song, Z.; Desmond, R.; Tschäen, D. M.; Grabowski, E. J. J.; Reider, P. J. *Tetrahedron Lett.* **1998**, *39*, 5323. (f) Lubell, W. D.; Jamison, T. F.; Rapoport, H. *J. Org. Chem.* **1990**, *55*, 3511. (g) Leanna, M. R.; Sowin, T. J.; Morton, H. E. *Tetrahedron Lett.* **1992**, *33*, 5029. (h) de Nooy, A. E. J.; Besemer, A. C.; van Bekkum, H. *Synthesis* **1996**, 1153. (i) Reddy, K. L.; Sharpless, K. B. *J. Am. Chem. Soc.* **1998**, *120*, 1207.

Scheme 2^a

^a (a) i. PPTS, MeOH, 24 h; ii. RuCl₃, H₅IO₆, CCl₄/CH₃CN; (b) i. TEMPO, NaOCl; ii. NaClO₂; (c) 1 M Jones reagent; (d) i. Dess–Martin/THF; ii. NaClO₂, NaH₂PO₄; (e) H₂, Pd/C.

17b, **20b**, and **30a** were oxidized efficiently with 1 N Jones reagent.^{14b} These conditions, however, afforded low yields in the cases of analogues protected with *N*-Boc or analogues containing activated aromatic rings such as **29a** and **12a/b**. Only analogues with alkyl-substituted aromatic groups were stable to the 1 N Jones reagent.

Dess–Martin oxidation of compound **29a** to the aldehyde stage followed by NaClO₂ treatment afforded the optically pure amino acid **38** in excellent yield. Compound **12b** after identical oxidation was fully deblocked to afford 4-methoxyhomophenylalanine (**40**) in excellent yield: [α]_D²⁰ –43.4 (*c* 0.1, 5 M HCl) [lit.^{3a} for (*S*)-**40** [α]_D²⁵ +42.2 (*c* 0.1, 5 M HCl)].

In summary, a concise method for the preparation of diverse α -amino alcohols and α -amino acids in both enantiomeric series from readily available achiral starting materials has been developed. Included in the analogues is 4-methoxyhomophenylalanine, a compound of current interest. The precursor vinyl amino alcohols and vinyl oxazo-

Table 3. Suzuki Coupling of (*R* and *S*)-3-(Benzyloxycarbonyl)-2,2-dimethyl-4-vinylloxazolines

| entry | Coupling partner | product | yield (%) |
|-------|------------------|---------|-----------------|
| 1 | | | 89 |
| 2 | | | 81 ^a |
| 3 | | | 71 ^a |
| 4 | | | 80 |
| 5 | | | 81 ^a |
| 6 | | | 77 |
| 7 | | | 73 ^a |
| 8 | | | 68 ^a |
| 9 | | | 66 |

^a HCl in MeOH, 24 h.

lidines were obtained via lipase-catalyzed kinetic resolutions. Key in this convergent approach was Pd-mediated Suzuki cross-coupling.

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Supporting Information Available: Experimental procedures and full characterization for all compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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